



PREDIS



Development of new radiochemical procedures for DTM radionuclides

SUB-TASK 4.5.3

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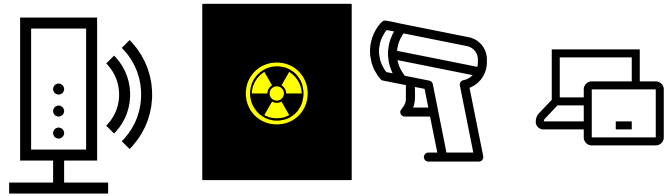
- Measurement and characterization of radionuclides difficult to measure DTM
 - Characterized by their β or X-rays emissions alone,
 - Associated to their low energy emission ($E < 100\text{keV}$) or low gamma intensity.
 - Difficulty to access to the standards calibrated with certificate to validate the process: rare, cost

Target	Reaction	DTM	Decay mode	ACTIVATED PRODUCTS
Li-6	(n, α)	H-3	β^-	
C-13	(n, α)	C-14	β^-	
Cl-35	(n, γ)	Cl-36	β^- -(β^+ , EC)	
Ca-40	(n, γ)	Ca-41	EC	
Fe-54	(n, γ)	Fe-55	EC	
Ni-58	(n, γ)	Ni-59	EC	
Ni-62	(n, γ)	Ni-63	β^-	
Zr-92	(n, γ)	Zr-93	β^-	
Mo-92	(n, γ)	Mo-93	EC	
Nb-93	(n, γ)	Nb-93m	IT	
Nb-93	(n, γ)	Nb-94	β^- , γ	
Mo-98	(n, γ)	Tc-99	β^-	
Sn-124	(n, γ)	Sb-125	β^- , γ	

DTM	Production mode	Decay mode	FISSION PRODUCTS
Sr-90	Fission	β^-	
Y-90	Fission, decay	β^-	
Tc-99	Fission, activation	β^-	
I-129	Fission	β^-	
Cs-135	Fission	β^-	

Evaluation of activities

Non destructive method



- Scaling factor to determine the activity level via the ratio with the activity of the RN that is easy to determine (Co-60, Cs-137)
- Gamma or neutron measurements for qualitative controls without sampling

– **Discrepancy** can exist between measured activities

- Corrosion of steels,
- Better knowledge of impurities in metallic materials

Destructive method

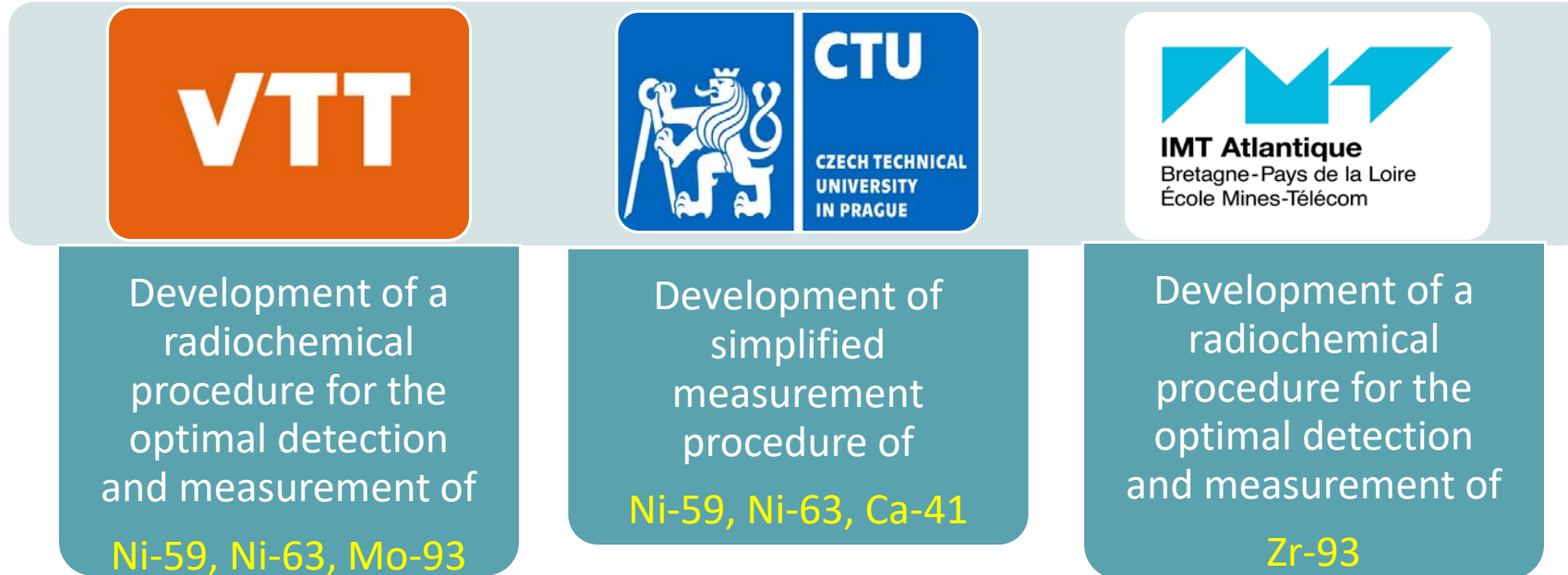


- To complete the non destructive method
- Important in particular when information is not accessible and available (chemical composition, irradiation conditions...) for the historical wastes



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Objectives



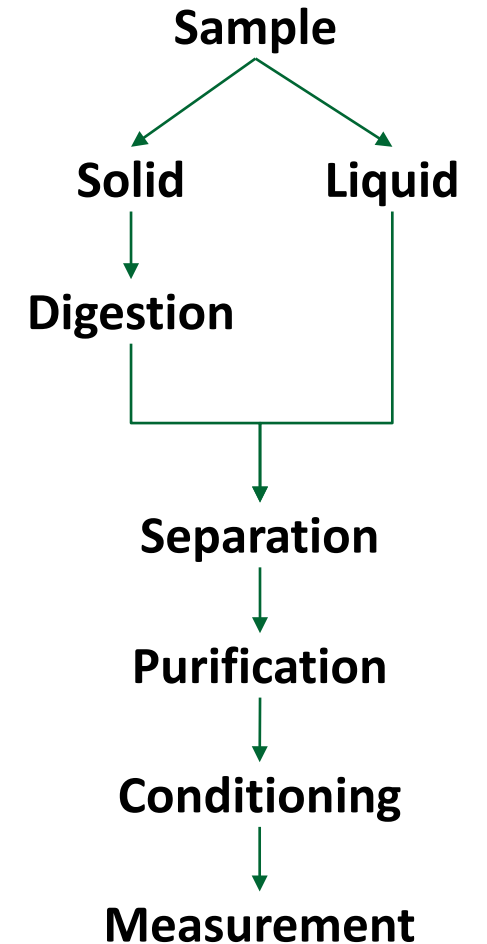
■ Targets:

- Highly selective and efficient separation and purification
- Development of sensitive to ultra-sensitive method of measurement, depending on the radionuclide

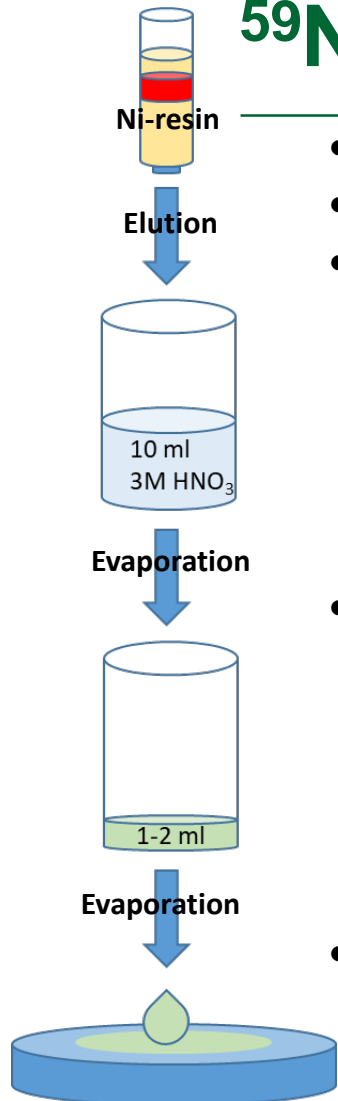


General methodology

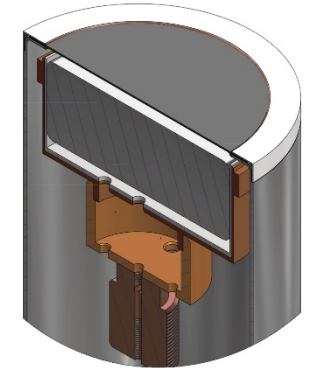
- **Sample**
 - Synthetic: prepared in laboratory from chemical reagents
 - From solid containing the radionuclide source
 - Surrogate to simulate the radionuclide due to limited quantity or availability
- **Separation / Purification**
 - Purification by chromatographic resins to recover the radionuclide of interest
 - Electrodeposition
- **Conditioning**
 - Sample preparation for measurements
 - Liquid form or deposition on filters or plates
- **Measurement**
 - Selection of an adapted analytical technique
 - Optimization of detection efficiency



^{59}Ni and ^{63}Ni analysis in activated steel



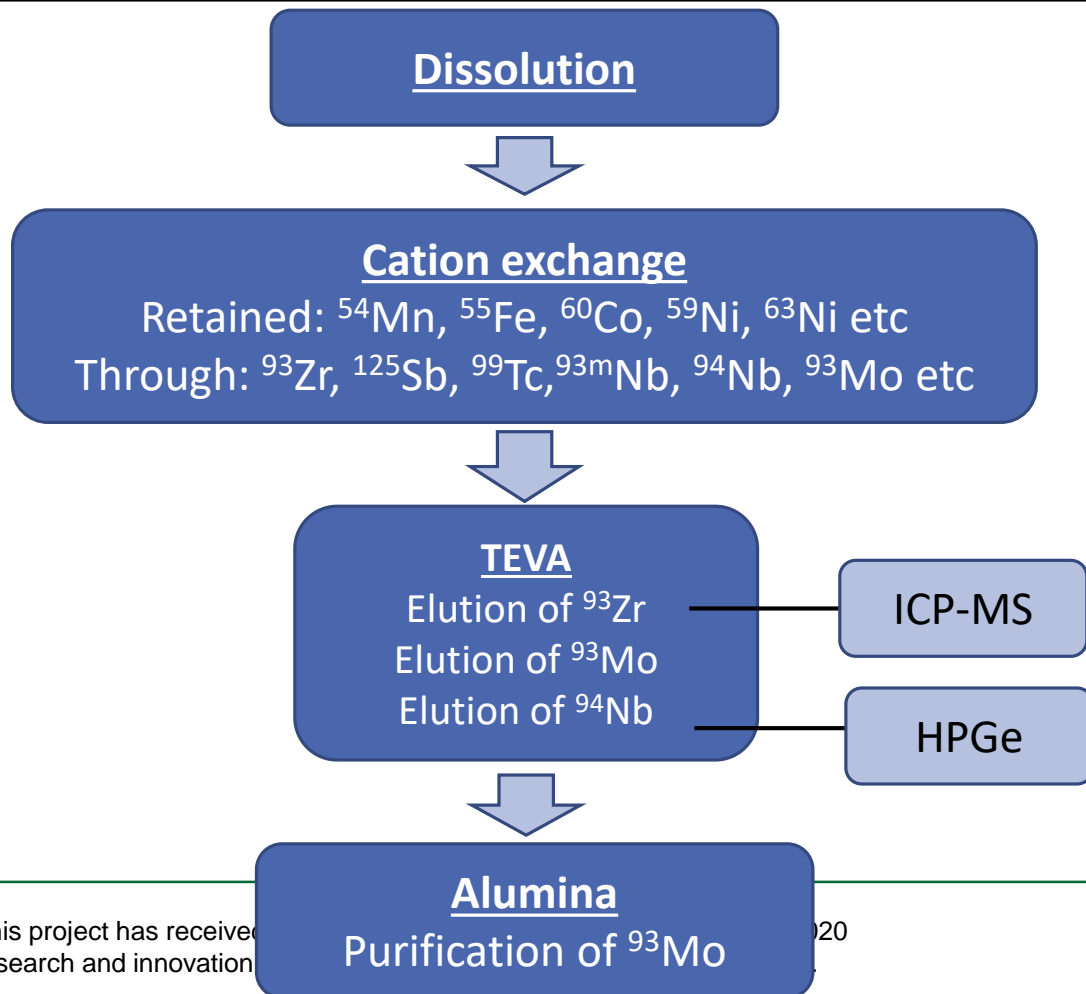
- ^{63}Ni analysis method published [1]
- Study to detect ^{59}Ni using a broad energy gamma spectrometer (3 keV to 3 MeV)
- Preparation of a standard curve
 - ^{55}Fe used as surrogate
 - Evaporation on a filter
 - Good linearity between 20-1000 Bq ($R^2 = 0,9991$)
 - When samples are measured, new data points can be added in the std curve prior the measurement i.e. verification that the curve is still valid
- Interference of ^{60}Co contamination
 - Co and Ni chemically similar \rightarrow ^{60}Co possible in purified Ni-fraction
 - ^{60}Co interference tested with 1000 Bq, 500 Bq and 100 Bq
 - Effect seen for ^{55}Fe activities below 250 Bq
 - Purity of Ni-fraction can be verified by gamma spectrometry prior to evaporation or during ^{59}Ni measurement as energy range covers also ^{60}Co
- Next step an intercomparison exercise and writing of a publication



^{93}Mo (^{94}Nb and ^{93}Zr) analysis

Sequential radiochemical purification of ^{93}Mo (^{94}Nb and ^{93}Zr)

Detection of ^{93}Mo using ICP-MS



Elimination of interferences

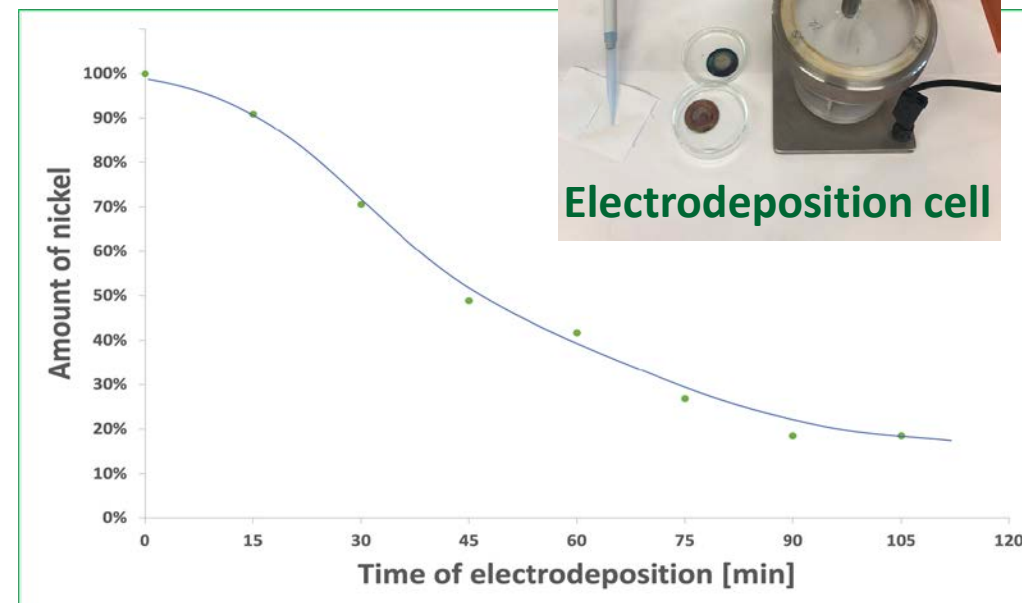
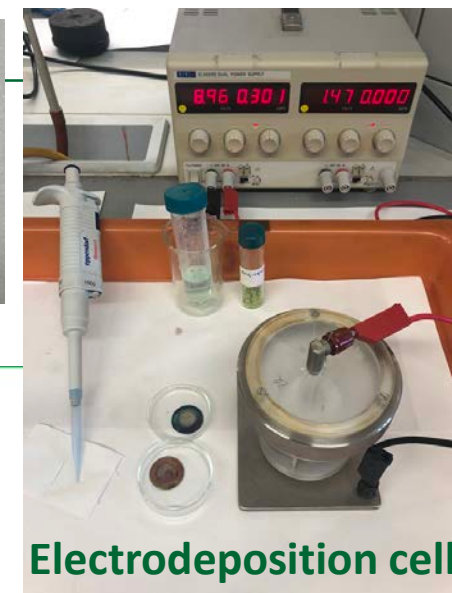
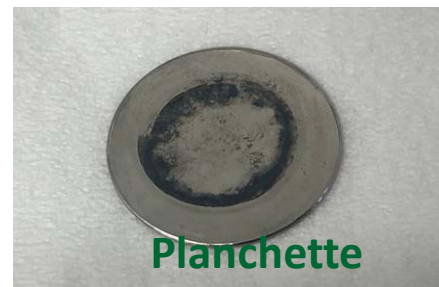
NEXT

Production of ^{93}Mo with proton bombardment of Nb



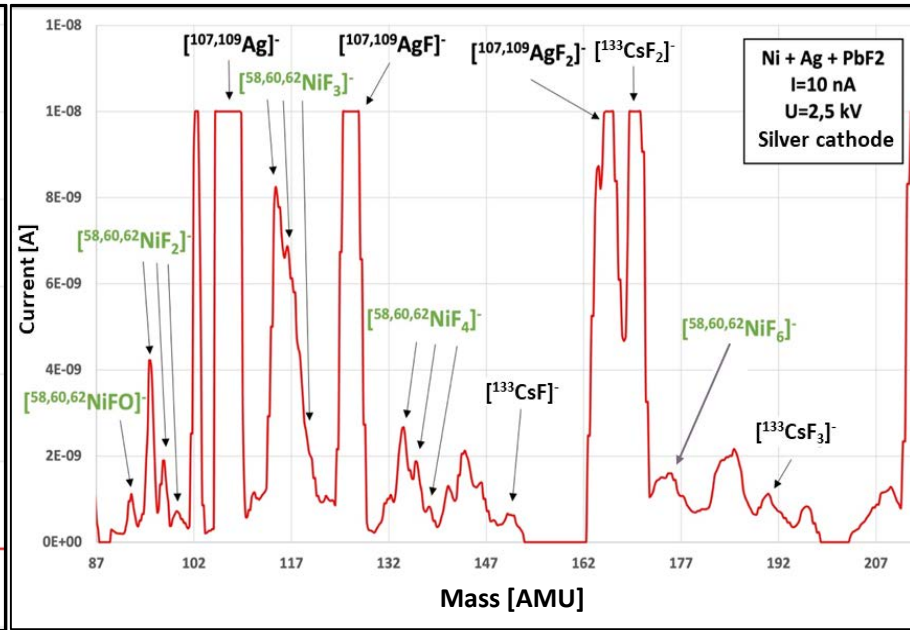
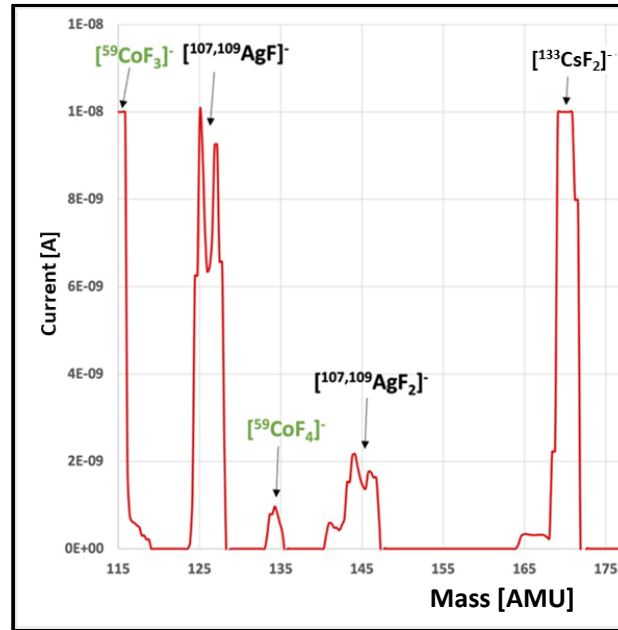
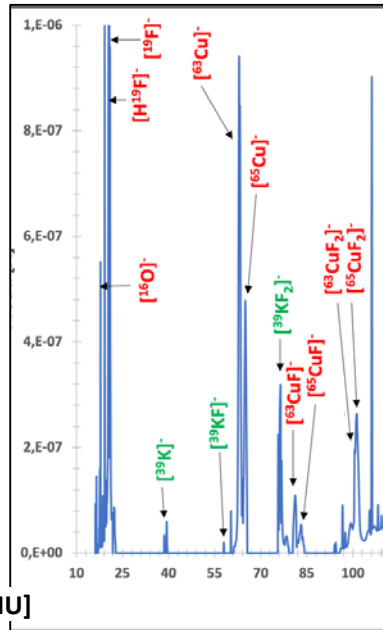
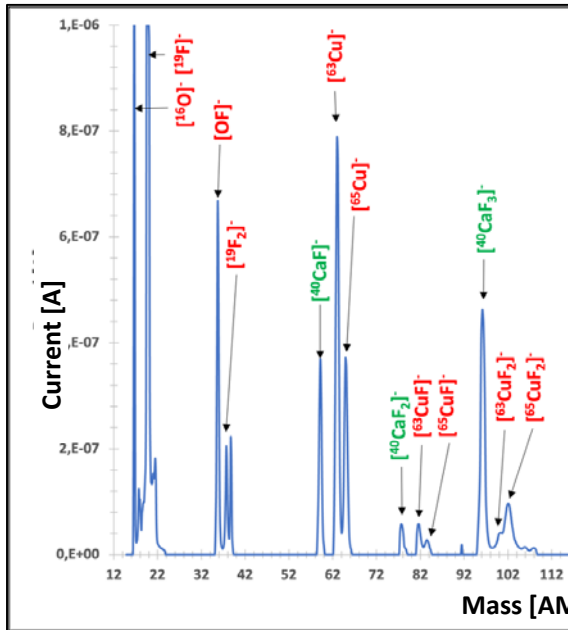
Ni isotopes:

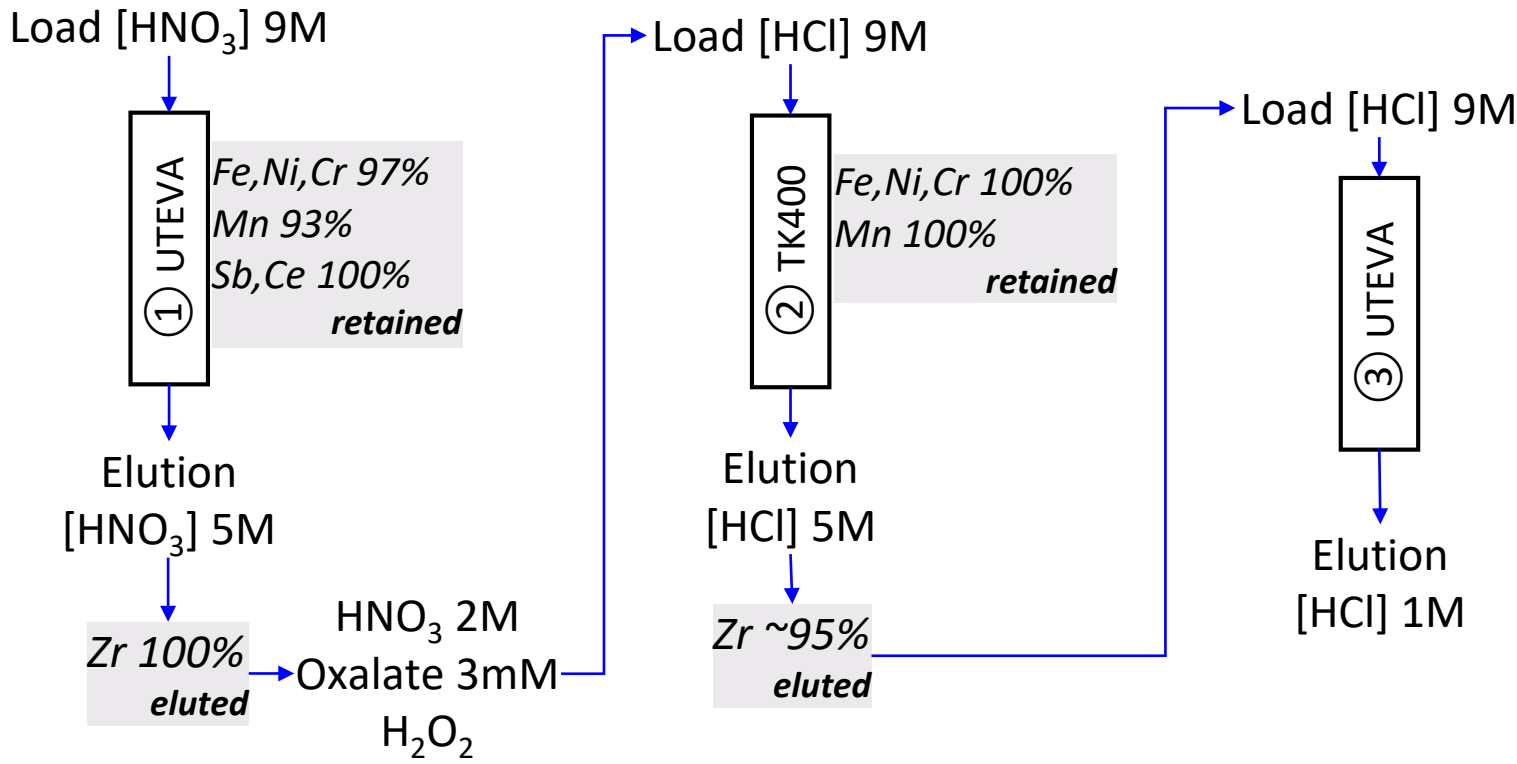
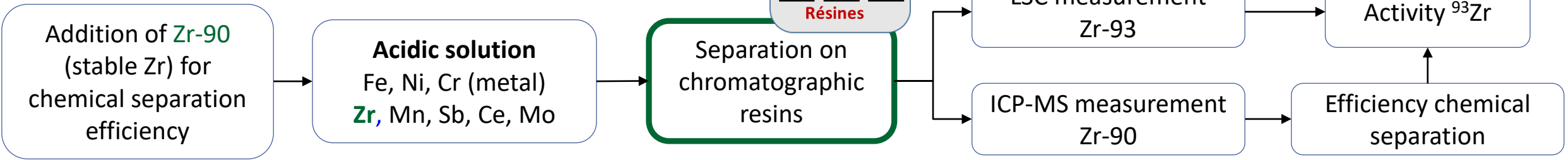
- How to improve measurement of $^{59,63}\text{Ni}$ and simplify the whole procedure?
 - Standard radiochemical separation using Ni-resin / DMG
 - Electrodeposition of Ni on a thin stainless steel planchette
 - Measure the planchette with the deposit towards detector in a calibrated geometry = ^{59}Ni
 - Place the planchette into a LSC vial with the deposit towards cocktail volume = ^{63}Ni
- Calibration needed (geometry) and LSC windows has to be adjusted
- Electrodeposition = additional separation and concentration step against non-depositing RN, higher sample volume can be used, carrier possible.
- Yield up to 90% over 2 hours of electrodeposition in H_3BO_3 solution at adjusted pH, 1.5 Acm^{-2}



Testing of suppression of ^{59}Ni and ^{41}Ca isobars for AMS measurement:

- Ultratrace amounts, suppression of stable ^{41}K and ^{59}Co needed.
- Mass scans to prove/find heavy negative fluoride molecules $[\text{MF}_n]^-$ - mass spectra
- Measured at caesium sputter ion source in the form of metals/salts mixed with PbF_2
- MS spectra for: ^{40}Ca ^{39}K ^{59}Co $^{58,60,62}\text{Ni}$



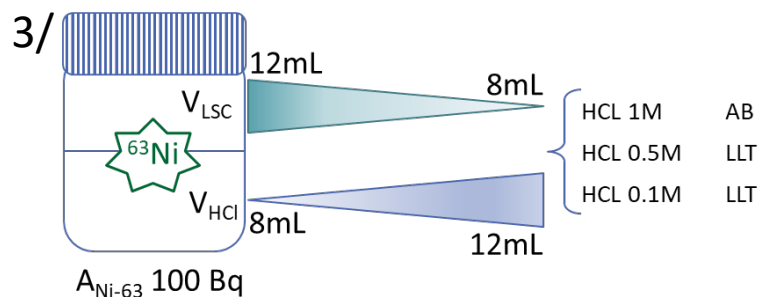


2g sample, 2h counting	background	Detection limit
HCl 1M / U.G. AB	0.07 cps	0.041 Bq/g
HCl 0.1M / U.G. AB	0.07 cps	0.039 Bq/g

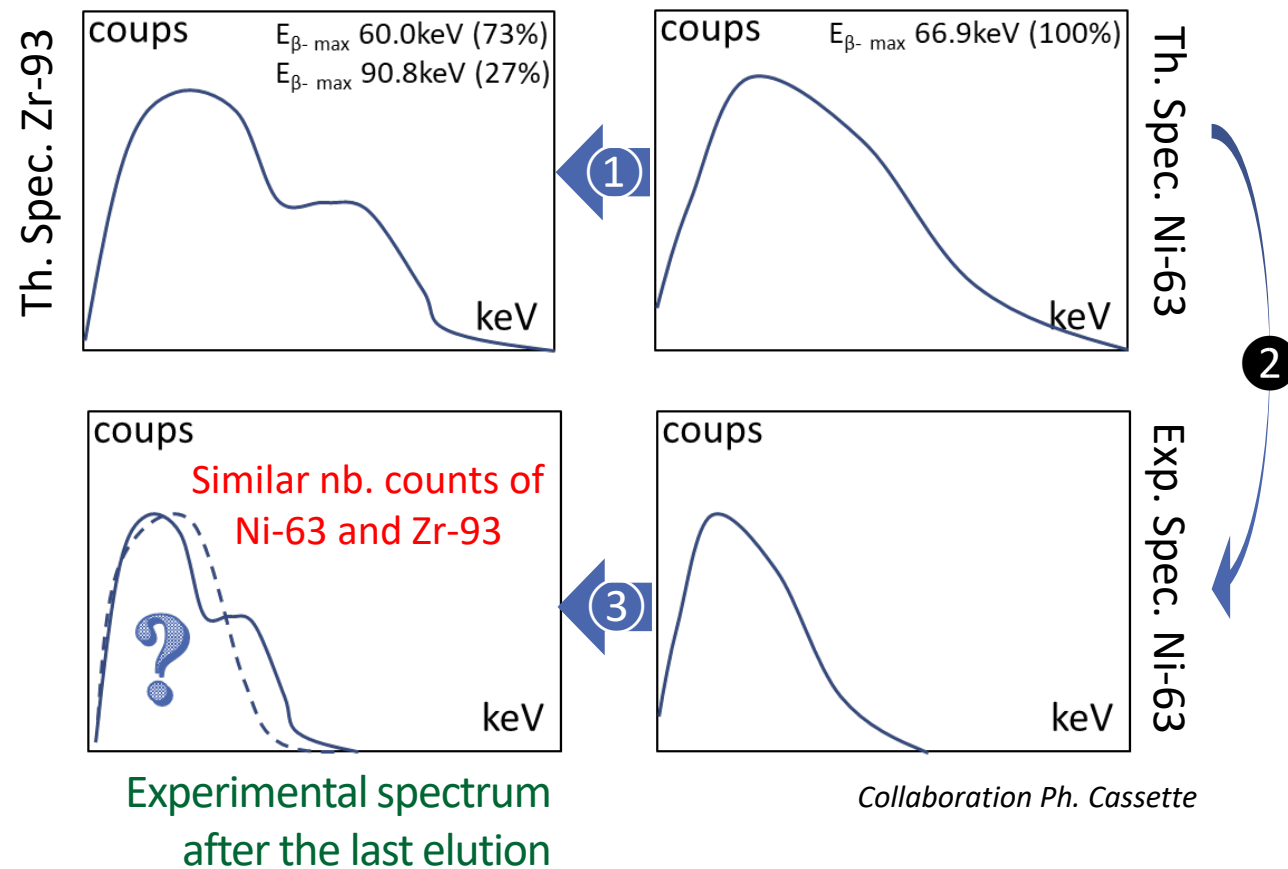
Zr-93 measurement

1/ **limited** supply of radioactive Zr-93 standard:
challenging temporal projection

2/ use of Ni-63 as a surrogate



4/ theoretical simulation of Ni-63 and Zr-93 spectra in the last elution condition for LSC analytical measurements conditions



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